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Discovery of cycloartane-type triterpene saponins from Mussaenda glabra

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ABSTRACT

Five new cycloartane saponins, mussaglaosides A–E (1–5) together with five known saponins, mussaendoside O, mussaendoside, G mussaendoside U, mussaendoside P, and mussaendoside Q (6–10) were isolated from the leaves of *Mussaenda glabra*. Their structures were determined on the basis of extensive spectroscopic methods, including 1D-, 2D-NMR, and MS data. All compounds were evaluated for ANO1 inhibitory activity using calcium-activated chloride channel and YFP expressing HT29 cells. Among the tested compounds, compound **6** strongly inhibited chloride channel activity with IC₅₀ value of 22.0 \pm 1.7 µM without any cytotoxicity.

1. Introduction

Anoctamin-1 (ANO1), a voltage-sensitive calcium-activated chloride channel, is highly expressed in human cancer cells and the gastrointestinal interstitial cells of Cajal. Interestingly, the ANO1 inhibitor decreases the proliferation of ANO1-overexpressed cell lines. Previous studies have demonstrated that inhibition of ANO1 may provide a new targeting approach for antitumor therapy and for inhibiting excessive smooth muscle contraction as well as diarrhea symptoms (Bill et al., 2014; Hwang et al., 2016). Furthermore, studies have been conducted to discover natural products (Namkung et al., 2010; Seo et al., 2016; Zhang et al., 2017) as novel ANO1 inhibitors due to their property of "metabolite-likeness", being biologically active, and acting as a substrate to deliver materials into intracellular sites of action in transporter systems (Harvey et al., 2015).

Mussaenda is a genus of flowering plants (Rubiaceae) that are an important source of natural medicine, consisting of 200 species over the world including 27 species in Vietnam. *Mussaenda* genus has been used in traditional medicine for the treatment of different ailments such as sore throat and stomach illness (Chi, 1999). Phytochemical studies of the *Mussaenda* species have revealed the presence of saponins, particularly unique *N*-triterpene saponins (Mohamed et al., 2016; Weimin et al., 1996) and iridoids (De et al., 2012; Dinda et al., 2008). In addition, these species have been shown to have anti-inflammatory,

antioxidant, and anticancer activities (Astalakshmi and Sundara, 2017). *Mussaenda pubescens* has been used for the treatment of diarrhea (Zhao et al., 1994). This paper reports the isolation, structural elucidation, and ANO1 inhibitory activity of five new and five known triterpenoid saponins from *Mussaenda glabra* Vahl (Fig. 1).

2. Results and discussion

Compound 1 was obtained as an amorphous powder and its molecular formula was determined to be $C_{60}H_{95}O_{23}N$ by HR-ESI-MS at m/z 1220.6189 $[M + Na]^+$ (Calcd. for $[C_{60}H_{95}O_{23}NNa]^+$, 1220.6187) see Supplement material. The ¹H-NMR spectrum of **1** (methanol- d_4) showed the signals of three olefin protons at $\delta_{\rm H}$ 6.04 (dd, J = 8.8, 14.8 Hz), 6.46 (dd, J = 10.8, 14.8 Hz), and 7.04 (d, J = 10.8 Hz), two protons of the cyclopropane ring at $\delta_{\rm H}$ 0.37 (br s) and 0.57 (br s); five methyl groups at $\delta_{\rm H}$ 0.88 (s), 0.93 (s), 1.04 (d, J = 6.5 Hz), 1.05 (s), and 1.07, (s) assigning to a cycloartane triterpene-type aglycone; four anomeric protons at $\delta_{\rm H}$ 4.41 (d, $J = 8.0 \,\text{Hz}$), 4.84 (br s), 4.90 (d, J = 7.6 Hz), and 5.23 (br s) suggesting the presence of four sugar units. In addition, two methyl groups at $\delta_{\rm H}$ 0.81 (d, J = 7.6 Hz) and 1.34 (d, J = 6.4 Hz), and one amino-methine proton at $\delta_{\rm H}$ 5.02 (d, 7.2 Hz) were assigned to a 2-amino-3-methyl-4-pentanolide (4-hydroxyisoleucine ylactone). The ¹³C-NMR and DEPT spectra (Table 1) revealed the signals of 60 carbons, including 8 non-protonated carbons, 31 methines, 12

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Fig. 1. Chemical structures of compounds 1-5.

methylenes, and 9 methyl carbons. The analysis of ¹H- and ¹³C-NMR data indicated the structure of 1 was similar to that of mussaendoside O (6) (Zhao et al., 1994) except for the addition of a hydroxyl group at C-26. The position of this group was proven by HMBC correlations between hydroxymethylene protons H-26 ($\delta_{\rm H}$ 4.43 and 4.47) and C-24 ($\delta_{\rm C}$ 139.5)/C-25 ($\delta_{\rm C}$ 130.6)/C-27 ($\delta_{\rm C}$ 171.1). The HMBC correlations from H₃-21 ($\delta_{\rm H}$ 1.04) to C-17 ($\delta_{\rm C}$ 53.1)/ C-20 ($\delta_{\rm C}$ 42.4)/C-22 ($\delta_{\rm C}$ 152.1); from H-24 ($\delta_{\rm H}$ 7.04) to C-22/C-23 ($\delta_{\rm C}$ 123.6)/C-25 ($\delta_{\rm C}$ 130.6)/C-26 ($\delta_{\rm C}$ 57.4)/C-27 ($\delta_{\rm C}$ 171.1); and from H-22 ($\delta_{\rm H}$ 6.04) to C-17/C-20/C-21 ($\delta_{\rm C}$ 20.1)/C-24 ($\delta_{\rm C}$ 139.5) confirmed the positions of two conjugated double bonds at C-22/C-23 and C-24/C-25 (Fig. 2). The configurations of these double bonds were proven as E by the large coupling constant of H-22 and H-23, J = 14.8 Hz, as well as by the NOESY correlations between H-23 ($\delta_{\rm H}$ 6.46) and H₂-26 ($\delta_{\rm H}$ 4.43 and 4.47). The HMBC correlations between hil 3-Me ($\delta_{\rm H}$ 0.81) and hil C-2 ($\delta_{\rm C}$ 56.1)/C-3 ($\delta_{\rm C}$ 39.4)/C-4 ($\delta_{\rm C}$ 78.8), hil 4-Me ($\delta_{\rm H}$ 1.34) and hil C-3/C-4, and the chemical shift of hil H-2 ($\delta_{\rm H}$ 5.02 (d, J = 7.2 Hz)/hil C-2 ($\delta_{\rm C}$ 56.1) and hil H-4 ($\delta_{\rm H}$ 4.74 (m))/C-4 ($\delta_{\rm C}$ 78.8) suggested the presence of a 4-hydroxyisoleucine y-lactone moiety. The observation of NOESY correlations (Fig. 3) between hil H-2 ($\delta_{\rm H}$ 5.02) and hil H-3 ($\delta_{\rm H}$ 2.83)/hil H-4 ($\delta_{\rm H}$ 4.74); between hil 3-Me ($\delta_{\rm H}$ 0.81) and 4-Me ($\delta_{\rm H}$ 1.34) and the ¹³C-NMR chemical shift of hil 3-Me ($\delta_{\rm C}$ 8.1) and hil 4-Me ($\delta_{\rm C}$ 15.7) confirmed the configuration of 4-hydroxyisoleucine γ -lactone as $[2S^*, 3R^*, 4R^*]$ by comparison with those of mussaendoside G [2S*,3R*,4R*] (Weimin et al., 1996) [hil 3-Me ($\delta_{\rm C}$ 8.0) and hil 4-Me ($\delta_{\rm C}$ 15.4)] and (2S,3R,4S)-4-hydroxyisoleucine γ -lactone hydrochloride [3-Me ($\delta_{\rm C}$ 13.0) and 4-Me $(\delta_{\rm C}$ 19.5)] (Wang et al., 2002). 4-hydroxyisoleucine γ -lactone connected with aglycone via the amide bond at C-27 was confirmed by HMBC correlations from hil H-2 ($\delta_{\rm H}$ 5.02) and C-27 ($\delta_{\rm C}$ 171.1), a specific amino acid lactone commonly found in the Mussaenda genus (Mohamed et al., 2016; Weimin et al., 1996). The HMBC correlations between H₃-28 ($\delta_{\rm H}$ 1.07)/H₃-29 ($\delta_{\rm H}$ 0.88) and C-3 ($\delta_{\rm C}$ 91.5) suggested the position of an oxygenated group at C-3. The a-configuration of H-3 (axial orientation) was determined by the observation of NOESY correlations between H-3 ($\delta_{\rm H}$ 3.23) and H-5 ($\delta_{\rm H}$ 1.33)/H₃-28 ($\delta_{\rm H}$ 1.07). Acid hydrolysis of 1 gave L-rhamnose and D-glucose (identified as trimethylsilyl (TMS) derivatives by gas chromatography-mass spectrometry (GC–MS)). In addition, the configurations of sugars in 1, β -D-glucopyranosyl and α -L-rhamnopyranosyl were supported by the multiplicity of glc I H-1 [$\delta_{\rm H}$ 4.41 (d, J = 8.0 Hz)], glc II H-1 [$\delta_{\rm H}$ 4.90 (d, J = 7.6 Hz)], rha I H-1 [$\delta_{\rm H}$ 5.32 (br s)], and rha II H-1 [$\delta_{\rm H}$ 4.84 (br s)]. The HMBC cross peaks from rha II H-1 ($\delta_{\rm H}$ 4.84) to glc I C-4 ($\delta_{\rm C}$ 79.7), from rha I H-1 ($\delta_{\rm H}$ 5.23) to glc II C-2 ($\delta_{\rm C}$ 78.9), from glc II H-1 ($\delta_{\rm H}$ 4.90) to glc I C-2 ($\delta_{\rm C}$ 78.5), and from glc I H-1 ($\delta_{\rm H}$ 4.41) to C-3 ($\delta_{\rm C}$ 91.5) confirmed the

sugar linkages to be α -L-rhamnopyranosyl- $(1\rightarrow 2)$ - $O-\beta$ -D-glucopyranosyl- $(1\rightarrow 2)$ - $[\alpha$ -L-rhamnopyranosyl $(1\rightarrow 4)$]- $O-\beta$ -D-glucopyranoside and the location of the sugar moiety at C-3 of the aglycone. This sugar moiety has been reported from *Mussaenda pubescens* (Zhao et al., 1994). Consequently, the structure of **1** was determined to be 26-hydro-xyheinsiagenin A 3- $O-\alpha$ -L-rhamnopyranosyl- $(1\rightarrow 2)-O-\beta$ -D-glucopyranosyl- $(1\rightarrow 4)$]- $O-\beta$ -D-glucopyranosyl- $(1\rightarrow 2)-[\alpha$ -L-rhamnopyranosyl- $(1\rightarrow 4)$]- $O-\beta$ -D-glucopyranoside and named mussaglaoside A.

Compound 2 was isolated as an amorphous powder and its molecular formula was determined to be $C_{60}H_{95}O_{24}N$ by HR-ESI-MS at m/z 1236.6134 $[M + Na]^+$ (Calcd. for $[C_{60}H_{95}O_{24}NNa]^+$, 1236.6136). The ¹H-NMR spectrum of **2** (methanol- d_4) exhibited the signals of a cycloartane triterpene-type aglycone and a 4-hydroxyisoleucine γ -lactone. In addition, four anomeric proton signals at $\delta_{\rm H}$ 5.03 (d, J = 7.2 Hz), 4.65 (d, J = 7.6 Hz), 4.59 (d, J = 7.2 Hz), and 4.37 (d, J = 8.0 Hz) were assigned to four sugar moieties. The ¹³C-NMR and DEPT spectra of 2 showed the presence of 60 carbons, including 8 non-protonated carbons, 31 methines, 13 methylenes, and 8 methyl carbons. The analytical ¹H and ¹³C NMR spectra of **2** indicated that the aglycone of **2** was similar to heinsiagenin A (Weimin et al., 1996). Acid hydrolysis of 2 gave D-glucose (identified as a TMS derivative). The sugar linkages were determined as β -D-glucopyranosyl- $(1 \rightarrow 2)$ -O- β -D-glucopyranosyl- $(1 \rightarrow 6)$ - $[O-\beta-D-glucopyranosyl-(1\rightarrow 2)]-O-\beta-D-glucopyranoside by the observa$ tion of HMBC correlations between glc IV H-1 ($\delta_{\rm H}$ 4.59) and glc III C-2 ($\delta_{\rm C}$ 83.2), glc III H-1 ($\delta_{\rm H}$ 5.03) and glc I C-6 ($\delta_{\rm C}$ 69.9), glc II H-1 ($\delta_{\rm H}$ 4.65) and glc I C-2 ($\delta_{\rm C}$ 79.4), and between glc I H-1 ($\delta_{\rm H}$ 4.37) and C-3 ($\delta_{\rm C}$ 91.8) and the location of a sugar at C-3 of the aglycone. Thus, compound **2** was determined as heinsiagenin A $3-O-\beta$ -D-glucopyranosyl- $(1\rightarrow 2)$ -*O*- β -D-glucopyranosyl- $(1\rightarrow 6)$ -[*O*- β -D-glucopyranosyl- $(1\rightarrow 2)$]-*O*- β -D-glucopyranoside and named mussaglaoside B.

The molecular formula of 3 was determined as C71H113O31N by the HR-ESI-MS at m/z 1498.7178 $[M + Na]^+$ (Calcd. for $[C_{71}H_{113}O_{31}NNa]^+$, 1498.7189). Similar to those of **2**, the aglycone of 3 was found to be heinsiagenin A (Weimin et al., 1996) by the analytical ¹H and ¹³C NMR data. Acid hydrolysis of 1 confirmed the sugar components as D-glucose, L-rhamnose, and D-xylose. In addition, the multiplicity of six anomeric protons [$\delta_{\rm H}$ 4.34 (d, J = 8.0 Hz, glc I H-1), 4.80 (d, J = 8.0 Hz, glc II H-1), 5.09 (d, J = 7.6 Hz, glc III H-1), 5.20 (br s, rha I H-1), 4.88 (br s, rha II H-1), and 4.52 (d, J = 7.2 Hz, xyl H-1)] suggested the presence of three β -D-glucopyranosyls, two L-rhamnopyranosyls, and one D-xylopyranosyl. Moreover, the HMBC correlations between xyl H-1 ($\delta_{\rm H}$ 4.52) and glc III C-2 ($\delta_{\rm C}$ 83.6), glc III H-1 ($\delta_{\rm H}$ 5.09) and glc II C-6 ($\delta_{\rm C}$ 69.9), rha II H-1 ($\delta_{\rm H}$ 4.88) and glc I C-4 ($\delta_{\rm C}$ 80.1), rha I H-1 ($\delta_{\rm H}$ 5.20) and glc II C-2 ($\delta_{\rm C}$ 79.2), and between glc II H-1 ($\delta_{\rm H}$ 4.80)

Table 1

(Astalakshmi and Sundara, 2017)H- and (Zhang et al., 2017)C-NMR spectroscopic data for compounds 1-5 in CD₃OD.

С	1		2		3		4		5	
	$\delta_{ m C}$	$\delta_{ m H}$ (mult, J in Hz)	$\delta_{ m C}$	$\delta_{ m H}$ (mult, J in Hz)	$\delta_{ m C}$	$\delta_{ m H}$ (mult, J in Hz)	$\delta_{ m C}$	$\delta_{ m H}$ (mult, J in Hz)	$\delta_{\rm C}$	$\delta_{ m H}$ (mult, J in Hz)
Aglycon										
1	33.1	1.27 (m)/1.55 (m)	33.2	1.25 (m)/1.52 (m)	33.2	1.25 (m)/1.50 (m)	40.5	1.46 (m)/1.57 (m)	33.2	1.24 (m)/1.54 (m)
2	30.3	1.73 (m)/2.04 (m)	30.9	1.69 (m)/2.04 (m)	30.9	1.71 (m)/2.03 (m)	70.6	3.75 (m)	30.4	1.72 (m)/2.05 (m)
3	91.5	3.23 (m)	91.8	3.14 (dd, 4.0, 11.6)	92.2	3.12 (dd, 4.0, 12.0)	96.8	2.97 (d, 8.8)	91.8	3.14 (dd, 4.0, 12.0)
4	42.2	-	42.3	-	42.4	-	43.2	-	42.3	-
5	48.4	1.33 (m)	49.0	1.30 (m)	48.4	1.30 (m)	48.7	1.34 (m)	48.4	1.30 (m)
6	22.1	0.80 (m)/1.59 (m)	22.2	0.83 (m)/1.63 (m)	22.2	0.83 (m)/1.59 (m)	22.1	0.83 (m)/1.58 (m)	22.2	0.80 (m)/1.59 (m)
/ Q	27.2 40.4	1.15 (III)/1.34 (III) 1.54 (m)	27.2 40.6	1.10 (III)/1.31 (III) 1.52 (m)	27.4 49.6	1.13 (III)/1.30 (III) 1.53 (m)	27.1 40 5	1.17 (III)/1.31 (III) 1.54 (m)	27.4 10.8	1.15 (III)/1.32 (III) 1.53 (m)
0 0	49.4 21.1	-	49.0 21.1	1.55 (iii)	49.0 21.1	1.55 (III) _	49.3 20.4	1.34 (III)	49.0 21.2	1.55 (III) _
10	21.1	_	21.1	_	21.1	_	26.4	_	21.2	_
11	27.5	1.15 (m)/2.04 (m)	27.4	1.16 (m)/2.02 (m)	27.5	1.16 (m)/2.03 (m)	27.6	1.17 (m)/2.05 (m)	27.5	1.15 (m)/2.02 (m)
12	34.0	1.68 (m)	34.1	1.67 (m)	34.1	1.67 (m)	34.0	1.69 (m)	34.1	1.65 (m)
13	46.7	_	46.7	_	46.8	-	46.7	-	46.9	-
14	50.1	_	50.1	-	50.1	-	50.1	-	49.5	-
15	36.8	1.30 (m)	36.8	1.30 (m)	36.8	1.30 (m)	36.8	1.30 (m)	36.8	1.34 (m)
16	29.4	1.30 (m)/1.74 (m)	29.5	1.30 (m)/1.74 (m)	29.5	1.28 (m)/1.73 (m)	29.5	1.30 (m)/1.74 (m)	28.4	1.37 (m)/1.83 (m)
17	53.1	1.74 (m)	53.2	1.73 (m)	53.2	1.73 (m)	53.1	1.73 (m)	50.5	1.68 (m)
18	19.0	1.05 (s)	18.9	1.05 (s)	18.9	1.05 (s)	19.0	1.05 (s)	18.5	1.02 (s)
19	30.9	0.37 (br s)/0.57 (br s)	30.7	0.36 (br s)/0.57 (br s)	30.9	0.36 (br s)/0.57 (br s)	30.8	0.47 (br s)/0.62 (br s)	30.9	0.35 (s)/0.56 (s)
20	42.4	2.28 (m)	42.5	2.27 (m)	42.5	2.26 (m)	42.2	2.26 (m)	43.9	1.73 (m)
21	20.1	1.04 (d, 6.5)	20.2	1.03 (d, 6.5)	20.2	1.03 (d, 6.5)	20.2	1.04 (d, 6.4)	12.5	0.93 (d, 6.8)
22	152.1	6.04 (dd, 8.8, 14.8)	149.7	5.92 (dd, 8.8, 14.8)	149.7	5.92 (dd, 8.8, 14.8)	149.6	5.93 (dd, 8.8, 14.8)	74.2	3.77 (m)
23	123.6	6.46 (dd, 10.8, 14.8)	124.5	6.34 (dd, 10.8, 14.8)	124.5	6.34 (dd, 10.8, 14.8)	124.6	6.34 (dd, 10.8, 14.8)	30.4	2.16 (m)/2.28 (m)
24 25	139.5 130.6	7.04 (d, 10.8) -	136.4 128.6	6.86 (d, 10.8) -	136.4 128.6	6.86 (d, 10.8) -	136.4 128.6	6.87 (d, 10.8) -	136.9 132.2	6.47 (t, 7.2) -
26	57.4	4.43 (d, 12.8) 4.47 (d, 12.8)	13.0	1.95 (s)	13.0	1.95 (s)	13.0	1.95 (s)	13.1	1.86 (s)
27	171.1	-	172.7	-	172.7	-	172.7	-	172.7	-
28	26.3	1.07 (s)	26.2	1.05 (s)	26.6	1.09 (s)	26.6	1.12 (s)	26.2	1.05 (s)
29	15.4	0.88 (s)	16.0	0.91 (s)	16.0	0.91 (s)	16.9	0.96 (s)	16.0	0.90 (s)
30	19.9	0.93 (s)	19.9	0.93 (s)	19.9	0.93 (s)	20.0	0.94 (s)	19.9	0.91 (s)
Hil										
1	176.8	-	176.9	-	176.9	-	176.8	-	177.1	-
2	56.1	5.02 (d, 7.2)	56.0	5.03 (d, 7.2)	56.0	5.03 (d, 7.2)	56.0	5.04 (d, 7.2)	53.8	4.55 (d, 9.6)
3	39.4	2.83 (m)	39.4	2.77 (m)	39.4	2.77 (m)	39.3	2.77 (m)	41.6	2.36 (m)
4	78.8	4.74 (m)	78.7	4.72 (m)	78.7	4.71 (m)	78.7	4.72 (m)	84.5	4.40 (m)
3-Me	7.9	0.81 (d, 7.6)	8.1	0.81 (d, 7.6)	8.1	0.85 (d, 7.6)	8.1	0.81 (d, 7.6)	12.4	0.93 (d, 7.2)
4-Me	15.7	1.34 (d, 6.4)	15./	1.33 (d, 6.4)	15.7	1.34 (d, 6.4)	15./	1.33 (d, 6.4)	19.9	1.40 (d, 6.0)
	Glc I		Glc I		Glc I		Glc I		Glc I	
1	105.3	4.41 (d, 8.0)	106.0	4.37 (d, 8.0)	106.2	4.34 (d, 8.0)	105.3	4.38 (d, 8.0)	106.0	4.36 (d, 8.0)
2	78.5	3.71 (m)	79.4	3.90 (m)	77.7	4.02 (m)	78.0	4.06 (m)	79.4	3.89 (m)
3	77.6	3.72 (m)	78.7	3.52 (m)	77.8	3.65 (m)	77.6	3.70 (m)	78.7	3.52 (m)
4	79.7	3.47 (m)	/1.1	3.42 (m)	80.1	3.60 (m)	79.4	3.65 (m)	71.1	3.42 (m)
5 6	70.4 62.0	3.32 (III) 3.65 (m) 3.79 (br.d. 11.2)	78.0 69.9	3.01 (III) 3.78 (dd 5.2, 11.2) 4.07 (br	70.4 62.2	3.28 (III) 3.68 (m)/3.78 (m)	70.7 62.6	3.41 (III) 3.68 (m)/3.80 (m)	78.0 69.9	3.01 (III) 3.78* 4.06 (br.d
0	02.0	5.65 (iii) 5.75 (bi d, 11.2)	09.9	d, 11.2)	02.2	0.00 (m)/ 0.7 0 (m)	02.0	0.00 (III)/ 0.00 (III)	05.5	11.2)
	Glc II		Glc II		Glc II		Glc II		Glc II	
1	102.0	4.90 (d, 7.6)	103.7	4.65 (d, 7.6)	101.9	4.80 (d, 8.0)	102.2	4.91 (d, 7.6)	103.7	4.65 (d, 7.6)
2	78.9	3.39 (m)	76.0	3.23 (m)	79.2	3.42 (m)	79.2	3.43 (m)	76.0	3.23 (m)
3	79.2	3.47 (m)	78.3	3.33 (m)	79.4	3.41 (m)	79.4	3.41 (m)	78.3	3.34 (m)
4	72.6	3.05 (t, 9.2)	71.8	3.20 (m)	72.0	3.08 (m)	72.6	3.11 (m)	71.8	3.20 (m)
5	78.0	3.24 (m)	78.4	3.28 (m)	77.5	3.50 (m)	77.6	3.50 (m)	78.4	3.27 (m)
0	63.7	d, 11.2)	62.4	3.69 (m)/3.80 (m)	69.9	3.70 (m)/4.05 (m)	/0.1	3.70 (m)/4.07 (m)	62.4	3.69 (m)/3.80 (m)
	Rha I		Glc III		Rha I		Rha I		Glc III	
1	101.7	5.23 (br s)	102.6	5.03 (d, 7.2)	101.9	5.20 (br s)	101.9	5.20 (br s)	102.6	5.03 (d, 7.2)
2	72.0	3.90 (br s)	83.2	3.39 (m)	72.3	3.89 (br s)	72.1	3.90 (br s)	83.2	3.39 (m)
3	72.2	3.73 (m)	77.5	3.22 (m)	72.2	3.72 (m)	72.2	3.72 (m)	77.6	3.22 (m)
4	74.1	3.39 (m)	71.4	3.34 (m)	74.2	3.38 (m)	74.1	3.38 (m)	71.4	3.33 (m)
5	09.4	4.11 (M) 1.22 (d. 6.4)	//.4 62.6	3.40 (m) 2.60 (m)/2.80 (m)	09.5 10 ₄	4.13 (M)	09.5	4.11 (M) 1.22 (d. 6.4)	/7.5	3.40 (M)
U	10.4	1.20 (U, U.4)	02.0	3.09 (III)/ 3.80 (III)	10.4	1.20 (U, 0.4)	10.4	1.22 (U, 0.4)	02.0	3.09 / 3.80"
	Rha II		Glc IV		Rha II		Rha II		Glc IV	
1	102.7	4.84 (br s)	105.6	4.59 (d, 7.2)	102.9	4.88 (br s)	102.8	4.87 (br s)	105.6	4.59 (d, 7.2)
2	72.3	3.83 (br s)	76.1	3.34 (m)	72.5	3.83 (br s)	72.4	3.83 (br s)	76.0	3.32 (m)
3	72.0	3.04 (m)	77.1	3.45 (m)	72.2	3.62 (m)	72.1	3.64 (m)	77.1	3.46 (m)
4	13.1	3.39 (M)	/1.3	3.29 (M)	/3.9	3.38 (M)	/3./	3.38 (m)	/1.4	3.29 (m)

(continued on next page)

Table 1 (continued)

С	1		2		3		4		5	
	$\delta_{ m C}$	$\delta_{ m H}$ (mult, J in Hz)	$\delta_{ m C}$	$\delta_{ m H}$ (mult, J in Hz)	$\delta_{ m C}$	δ_{H} (mult, J in Hz)	$\delta_{ m C}$	$\delta_{ m H}$ (mult, J in Hz)	$\delta_{ m C}$	$\delta_{ m H}$ (mult, J in Hz)
5	70.6	3.97 (m)	77.7	3.28 (m)	70.8	3.94 (m)	70.6	3.96 (m)	77.8	3.28 (m)
6	18.1	1.27 (d, 6.4)	62.7	3.69 (m)/3.80 (m)	18.3	1.30 (d, 6.4)	18.1	1.28 (d, 6.4)	62.7	3.69 (m)/3.80 (m)
					Glc III		Glc III			
1					102.7	5.09 (d, 7.6)	104.6	4.70 (d, 8.0)		
2					83.6	3.32 (m)	75.2	3.17 (m)		
3					78.1	3.34 (m)	77.6	3.70 (m)		
4					71.4	3.34 (m)	71.6	3.30 (m)		
5					77.7	3.65 (m)	77.8	3.33 (m)		
6					62.4	3.68*/3.78*	62.7	3.52 (m)/3.83 (m)		
					Xyl					
1					106.5	4.52 (d, 7.2)				
2					75.7	3.38 (m)				
3					77.4	3.30 (m)				
4					71.0	3.55 (m)				
5					67.2	3.19 (m)/3.94 (m)				

Assignments were done by HSQC, HMBC, and COSY experiments; *overlapped signal; Glc, glucopyranosyl; Rha, rhamnopyranosyl; Xyl, xylopyranosyl; Hil, 4-hydroxyisoleucine lactone.



Fig. 2. The key HMBC and COSY correlations of compounds 1-5.

and glc I C-2 ($\delta_{\rm C}$ 77.7) indicated the sequence of sugar linkages to be β -D-xylopyranosyl-(1 \rightarrow 2)-O- β -D-glucopyranosyl-(1 \rightarrow 2)-O-[α -L-rhamnopyranosyl-(1 \rightarrow 2)]-O- β -D-glucopyranosyl-(1 \rightarrow 2)-O-[α -L-rhamnopyranosyl-(1 \rightarrow 4)]-O- β -D-glucopyranoside. The positions of the sugar linkages at C-3 was confirmed by the HMBC correlations from glc I H-1 ($\delta_{\rm H}$ 4.34) to C-3 ($\delta_{\rm C}$ 92.2). Based on the above evidence, the structure of **3** was elucidated as heinsiagenin A 3-O- β -D-xylopyranosyl-(1 \rightarrow 2)-O- β -D-glucopyranosyl-(1 \rightarrow 2)-O- β -D-glucopyranosyl-(1 \rightarrow 2)-O- β -D-glucopyranosyl-(1 \rightarrow 2)-O- β -D-glucopyranosyl-(1 \rightarrow 2)-O-[α -L-rhamnopyranosyl-(1 \rightarrow 4)]-O- β -D-glucopyranosyl-(1 \rightarrow 2)-O-[α -L-rhamnopyranosyl-(1 \rightarrow 4)]-O- β -D-glucopyranosyl-(1 \rightarrow 2)-O-[α -L-rhamnopyranosyl-(1 \rightarrow 4)]-O- β -D-glucopyranosyl-(1 \rightarrow 4)]-O- β -D-glucopyranos

The molecular formula of **4** was determined as $C_{66}H_{105}O_{28}N$ by the HR-ESI-MS at m/z 1382.6703 $[M + Na]^+$ (Calcd. for

 $[C_{66}H_{105}O_{28}NNa]^+$, 1382.6715). The NMR data of 4 were similar to those of mussaendoside P (Zhao et al., 1994) except for the addition of one sugar unit at Glc II C-6. The HMBC correlations between H₃-28 (δ_{H} 1.12)/H₃-29 (δ_{H} 0.96) and C-3 (δ_{C} 96.8)/C-4 (δ_{C} 43.2)/C-5 (δ_{C} 48.7) as well as COSY correlations between H-2 (δ_{H} 3.75) and H-3 (δ_{H} 2.97) indicated the position of a hydroxyl group at C-2 and an oxygenated group at C-3. The large coupling constant of H-2 and H-3, J = 8.8 Hz and the NOESY correlations between H-2 (δ_{H} 3.75) and H₃-29 (δ_{H} 0.96)/H_a-19 (δ_{H} 0.47) indicated that the orientations of H-2 and H-3 were *axial*. The HMBC correlations from H₃-21 (δ_{H} 1.04) to C-17 (δ_{C} 53.1)/C-20 (δ_{C} 42.2)/C-22 (δ_{C} 149.6), H-24 (δ_{H} 6.87) to C-22/C-23 (δ_{C} 124.6)/C-25 (δ_{C} 128.6)/C-26 (δ_{C} 13.0)/C-27 (δ_{C} 172.7), from H-22 (δ_{H}



Fig. 3. The key NOESY correlations of compounds 1-5.

5.93) to C-17/ C-20/C-21 ($\delta_{\rm C}$ 20.2)/C-24 ($\delta_{\rm C}$ 136.4) confirmed the positions of two double bonds at C-22/C-23 and C-24/C-25. In addition, the large coupling constant of H-22 and H-23, J = 14.8 Hz and the NOESY correlations between H-23 ($\delta_{\rm H}$ 6.34) and H₃-26 ($\delta_{\rm H}$ 1.95) confirmed the configurations of both double bonds as *E*. Acid hydrolysis of 4 gave L-rhamnose and D-glucose (identified as TMS derivatives by GC-MS) GC-MS). The HMBC correlations between rha II H-1 ($\delta_{\rm H}$ 4.87) to glc I C-4 ($\delta_{\rm C}$ 79.4), rha I H-1 ($\delta_{\rm H}$ 5.20) and glc II C-2 ($\delta_{\rm C}$ 79.2), glc III H-1 ($\delta_{\rm H}$ 4.70) and glc II C-6 ($\delta_{\rm C}$ 70.1), glc II H-1 ($\delta_{\rm H}$ 4.91) and glc I C-2 ($\delta_{\rm C}$ 78.0), and between glc I H-1 ($\delta_{\rm H}$ 4.38) and C-3 ($\delta_{\rm C}$ 96.8) indicated the sequence of sugar moieties. Thus, the structure of 4 was determined as 2 α -hydroxyheinsiagenin A 3-O- β -D-glucopyranosyl-(1 \rightarrow 6)-O-[α -L-rhamnopyranosyl-(1 \rightarrow 4)]-O- β -D-glucopyranoside and named as mussaglaoside D.

The HR-ESI-MS of 5 corresponded to the molecular formula of C₆₀H₉₇O₂₅N. The ¹H and ¹³C-NMR spectral data of **5** were similar to those of 2 except for the difference branch chain of cycloartane. In addition, the position of a hydroxyl group at C-22 was proved by HMBC correlations between H_3-21 ($\delta_{\rm H}$ 0.93) and C-17 ($\delta_{\rm C}$ 50.5)/C-20 ($\delta_{\rm C}$ 43.9)/C-22 ($\delta_{\rm C}$ 74.2). The α -configuration of the hydroxyl group at C-22 was proven by NOESY correlations between H-22 ($\delta_{\rm H}$ 3.77) and H-20 $(\delta_{\rm H} 1.73)$ /H-16 ($\delta_{\rm H} 1.37$ and 1.83). The 4-hydroxyisoleucine γ -lactone connected with cycloartane aglycone via the N-amide bond which was confirmed by HMBC correlations from hil H-2 ($\delta_{\rm H}$ 4.55) to C-27 ($\delta_{\rm C}$ 172.7). The $^{13}\text{C-NMR}$ chemical shift of hil 3-Me (δ_{C} 12.4) and hil 4-Me ($\delta_{\rm C}$ 19.9) confirmed the configuration of 4-hydroxyisoleucine γ -lactone $(2S^*, 3R^*, 4S^*)$ by comparison with those of mussaglaosides B-D (2-4) [hil 3-Me (δ_C 8.1) and hil 4-Me (δ_C 15.7)] and (2S,3R,4S)-4-hydroxyisoleucine γ -lactone hydrochloride [3-Me ($\delta_{\rm C}$ 13.0) and 4-Me ($\delta_{\rm C}$ 19.5)] (Wang et al., 2002), and were further confirmed by NOESY correlations between hil 4-Me ($\delta_{\rm H}$ 1.40) and hil H-2 ($\delta_{\rm H}$ 4.55)/hil H-3 ($\delta_{\rm H}$ 2.36). The sugar linkages of **5** were proven to be similar to those of mussaglaoside B (2). Thus, compound 5 was determined as N-(2S*,3R*,4S*)-hydroxyisoleucine γ-lactone-22-hydroxycycloart-24(25)ene-27-amide 3-*O*- β -D-glucopyranosyl-(1 \rightarrow 2)-*O*- β -D-glucopyranosyl- $(1\rightarrow 6)$ -O- $[\beta$ -D-glucopyranosyl- $(1\rightarrow 2)$]-O- β -D-glucopyranoside and named mussaglaoside E.

The known compounds (6-10) were characterized as mussaendoside O (Zhao et al., 1994), mussaendoside G (Weimin et al., 1996) mussaendoside U (Zhao et al., 1997), mussaendoside P (Zhao et al., 1994), and mussaendoside Q (Zhao et al., 1994), respectively by analysis of NMR spectroscopic data and in comparison in the literature.

To determine the inhibitory effect of ANO1 channel activity on HT29 human colon cancer cell lines, yellow fluorescent protein (YFP) reduction assays were performed using triterpenoid saponins (1–10) isolated from *M. glabra*. ANO1 chloride channel activity was measured using HT29 (human colon cancer cells) endogeneously expressing

human ANO1 and the genetically encoded iodide-sensing fluorescent protein, YFP-F46 L/H148Q/I152 L (Seo et al., 2016). HT-29 cells were pre-incubated with test compounds in PBS prior to the addition of a solution containing iodide and ATP. In the presence of an ANO1 inhibitor, YFP fluorescence quenching by iodide intake through ANO1 was inhibited. As shown in Table 2, compounds 2–6, and 10 blocked iodide influx by > 25% at the concentration of 30 μ M. Compound 6 significantly inhibited ANO1 chloride channel activity by 53% without any cytotoxicity. Compound 6 was further evaluated at various concentrations to get IC₅₀ value. As the results, compounds 6 strongly inhibited chloride channel activity with an IC₅₀ value of 22.0 ± 1.7 μ M. These data suggest that triterpenes from the leaves of *M. glabra* may be powerful candidates in the development of novel therapeutic antitumor and inhibitory agents for excessive smooth muscle contraction and diarrhea symptoms *via* ANO1 inhibition.

3. Experimental

3.1. General

All NMR spectra were recorded on an AGILENT 400 MHz. Data processing was carried out with the MestReNova ver. 9.0.1. Optical rotations were determined on a Jasco DIP-370 digital polarimeter. HR-ESI-MS spectra were obtained using AGILENT 6550 iFunnel Q-TOF LC/MS system. HPLC was carried out using an AGILENT 1200 HPLC system. Column chromatography (CC) was performed on silica-gel (Kieselgel 60, 230–400 mesh, Merck) or RP-18 resins (30–50 μ m, Fuji Silysia Chemical Ltd.). For thin layer chromatography (TLC), pre-coated silica-gel 60 F₂₅₄ (0.25 mm, Merck) and RP-18 F₂₅₄S (0.25 mm, Merck) plates were used.

Table 2
PAR2 receptor inhibition of compounds 1-10 at 30 μM in HT29-YFP
cells.

Compounds	PAR2 receptor inhibition (%)				
1	< 25				
2	30.8 ± 1.5				
3	42.7 ± 3.1				
4	33.0 ± 2.7				
5	34.8 ± 4.5				
6	53.4 ± 3.4				
7	< 25				
8	< 25				
9	37.2 ± 2.5				
10	< 25				

3.2. Plant material

The aerial parts of *Mussaenda glabra* Vahl were collected at Me Linh, Vinh Phuc, Vietnam in February 2017 and identified by Dr. Nguyen The Cuong, Institute of Ecology and Biological Resources, VAST. A voucher specimen (NCCT-P67) was deposited at the Institute of Marine Biochemistry, VAST.

3.3. Extraction and isolation

The dried powder of aerial parts of *M. glabra* (5.2 kg) was sonicated 3 times with hot methanol and then filtered through filter paper, removed solvent to yield 270 g of a dark solid extract. The extract was suspended in water and successively partitioned with n-hexane, dichloromethane, and ethyl acetate solvents giving *n*-hexane (MG1, 45 g), dichloromethane (MG2, 90 g), ethyl acetate extracts (MG3, 23 g) and water layer (MG4). Continuously, the water layer was chromatographed on a Diaion HP-20 column eluting with water to remove sugar components, then increase concentration of methanol in water (50 and 100%, v/v) to obtain two fractions, MG4A and MG4B, respectively. The MG4A fraction was chromatographed on a silica gel column eluting with dichloromethane/methanol (20/1, 10/1, 5/1, 2.5/1, v/v) to give four sub-fractions MG4A1-MG4A4. MG4A3 was chromatographed on a RP-18 column eluting with acetone/water (1/1.2, v/v) to give four smaller fractions, MG4A3A-MG4A3D. MG4A3C fraction was chromatographed on a silica gel column eluting with dichloromethane/methanol/water (3/1/0.1, v/v/v) to give two fractions, MG4A3C1-MG4A3C2. MG4A3C1 further chromatographed on HPLC using J'sphere ODS M-80, 150 mm \times 20 mm column, ACN in H2O (38%, v/v) and a flow rate of 3 mL/min to yield compounds 4 (8.0 mg) and 10 (40.0 mg). Compound 5 (6.0 mg) was obtained from MG4A3C2 chromatographed on HPLC using J'sphere ODS M-80, $150 \text{ mm} \times 20 \text{ mm}$ column, ACN in H₂O (35%, v/v), and a flow rate of 3 mL/min. The MG4B fraction was separated into four smaller fraction MG4B1-MG4B4 by silica gel column and gradient solvent system of dichloromethane/ methanol (20/1, 10/1, 5/1, 2.5/1, v/v). MG4B2 was chromatographed on a RP-18 column eluting with acetone/water (1/1, v/v) to give three smaller fractions, MG4B2A-MG4B2C. MG4B2A fraction was chromatographed on a silica gel column eluting with dichloromethane/ acetone/water (1/4/0.3, v/v/v) to give two fractions, MG4B2A1-MG4B2A2. The MG4B2A1 was purified on HPLC column (J'sphere ODS M-80, 150 mm \times 20 mm column) eluting with 45% acetonitrile to yield 1 (20.0 mg). MG4B2A2 chromatographed on HPLC using J'sphere ODS M-80, 150 mm \times 20 mm column, 45% CAN in H₂O to yield 2 (6.0 mg). MG4B2C fraction was chromatographed on a silica gel column eluting with dichloromethane/methanol/water (3/1/0.1, v/v/v) to give two fractions, MG4B2C1 and MG4B2C2. MG4B2C2 was further chromatographed on HPLC using J'sphere ODS M-80, 150 mm \times 20 mm column, ACN in H_2O (45%) and a flow rate of 3 mL/min to yield 6 (60.0 mg). MG4B4 was chromatographed on a RP-18 column eluting with acetone/ water (1/1.2, v/v) to give five smaller fractions, MG4B4A-MG4B4E. MG4B4A fraction was chromatographed on a silica gel column eluting with dichloromethane/acetone/water (1/4/0.3, v/v/v) to give two fractions, MG4B4A1-MG4B4A2. MG4B4A2 was further chromatographed on HPLC using J'sphere ODS M-80, 150 mm \times 20 mm column, ACN in H₂O (45%) to yield 9 (30.0 mg). MG4B4C fraction was chromatographed on a silica gel column eluting with dichloromethane/ methanol/water (3/1/0.1, v/v/v) to give two fractions, MG4B4C1-MG4B4C2. The MG4B4C1 was purified on HPLC column (J'sphere, ODS M-80, 150 mm \times 20 mm) eluting with 40% acetonitrile to yield compounds 8 (40.0 mg) and 7 (7.0 mg). Finally, compound 3 (15.0 mg) was isolated from the MG4B4C2 chromatographed on HPLC using J'sphere ODS M-80, 150 mm \times 20 mm column, ACN-H₂O (40–60, v/v).

3.3.1. Mussaglaoside A (1)

White amorphous powder. $[\alpha]_D^{25} = +30.0$ (c 1.0 MeOH). HR-ESI-

MS m/z: 1220.6189 $[M + Na]^+$ (Calcd. for $[C_{60}H_{95}O_{23}NNa]^+$, 1220.6187). ¹H- and ¹³C-NMR (CD₃OD): see Table 1.

3.3.2. Mussaglaoside B (2)

White amorphous powder. $[\alpha]_D^{25} = +45.0$ (*c* 1.0 MeOH). HR-ESI-MS m/z: 1236.6134 [M + Na]⁺ (Calcd. for $[C_{60}H_{95}O_{24}NNa]^+$, 1236.6136). ¹H- and ¹³C-NMR (CD₃OD): see Table 1.

3.3.3. Mussaglaoside C (3)

White amorphous powder. $[\alpha]_D^{25} = +39.0$ (*c* 1.0 MeOH). HR-ESI-MS m/z: 1498.7178 [M + Na]⁺ (Calcd. for $[C_{71}H_{113}O_{31}NNa]^+$, 1498.7189). ¹H- and ¹³C-NMR (CD₃OD): see Table 1.

3.3.4. Mussaglaoside D (4)

White amorphous powder. $[\alpha]_D^{25} = +52.0$ (*c* 1.0 MeOH). HR-ESI-MS m/z: 1382.6703 [M + Na]⁺ (Calcd. for $[C_{66}H_{105}O_{28}NNa]^+$, 1382.6715). ¹H- and ¹³C-NMR (CD₃OD): see Table 1.

3.3.5. Mussaglaoside E (5)

White amorphous powder. $[\alpha]_D^{25} = +21.0$ (*c* 1.0 MeOH). HR-ESI-MS m/z: 1254.6248 [M + Na]⁺ (Calcd. for $[C_{60}H_{97}O_{25}NNa]^+$, 1254.6242). ¹H- and ¹³C-NMR (CD₃OD): see Table 1.

3.4. Acid hydrolysis

Each compound (1-5, 2.0 mg) was separately dissolved in 1.0 N HCl (dioxane-H₂O, 1:1, v/v, 1.0 mL) and heated to 80 °C in a water bath for 3 h. Acidic solution was neutralized with Ag₂CO₃ with the solvent thoroughly removed under a N₂ stream overnight. After extraction with CHCl₃, the aqueous layer was concentrated to drvness using N₂. The residue was dissolved in dry pyridine (0.1 mL), followed by addition of L-cysteine methyl ester hydrochloride in pyridine (0.06 M, 0.1 mL). The reaction mixture was heated at 60 °C for 2 h. Trimethylsilylimidazole solution (0.1 mL) was then added, followed by heating at 60 °C for 1.5 h. The dried product was partitioned with n-hexane and H₂O (0.1 mL each), and the organic layer was analyzed by gas chromatography (GC): column DB-5 (0.32 mm ID \times 30 m length), detector FID, column temp 210 °C, injector temp 270 °C, detector temp 300 °C, carrier gas He (2 mL/min). Under these conditions, the standard sugars gave peaks at t_R (min) 14.11 and 14.26 for D- and L-glucose, 8.21 and 8.66 for D- and L-xylose, and 4.50 for L-rhamnose, respectively. Peaks at t_R (min) 14.11 and 4.50 of D-glucose and L-rhamnose for 1 and 4; 14.11 of Dglucose for 2 and 5; 14.11, 8.21, and 4.50 of D-glucose, D-xylose, and Lrhamnose for 3 were observed.

3.5. Yellow fluorescent protein (YFP) fluorescence measurement

Calcium-activated chloride channel (ANO1) and YFP expressing HT29 cells were plated in 96-well black-walled microplates (Corning Inc., Corning, NY) at a density of 15,000 cells per well. YFP reduction analysis was perforemed usnig FLUOstar Omega microplate reader (BMG Labtech, Ortenberg, Germany) and MARS Data Analysis Software (BMG Labtech) as described in previous study (Seo et al., 2016). Briefly, each well of 96-well plate was washed 2 times in PBS (200 µL/wash) and test compounds were added to each well at 30 µM final concentration. After incubation at 37 °C for 10 min, 96-well plates were transferred to a plate reader for fluorescence analysis. Each well was individually assayed for ANO1-mediated I⁻ influx by monitoring YFP fluorescence continuously (400 ms per point) for 2s (baseline), then 140 mM I⁻ solution containing 200 µM ATP was added at 2 s and then YFP fluorescence was recorded for 6 s. The initial iodide influx rate was determined from the initial slope of fluorescence reduction by nonlinear regression after iodide injection with ATP.

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Appendix A. Supplementary data

Supplementary material related to this article can be found, in the online version, at doi:https://doi.org/10.1016/j.phytol.2019.07.006.

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